

1. Mo k-series radiation has an absorption edge corresponding to a wavelength of 0.61977 Å. The $K\alpha_1$ line has a wavelength of 0.70926 Å and the $K\beta_1$, a wavelength of 0.63225 Å.

A. What is the minimum potential in KV that can be used to produce Mo k-series radiation from a Mo-target X-ray tube?

Convert the energy of the absorption edge to eV.

$$E = hc/\lambda$$

$$E = 6.6 \times 10^{-34} \times 3.0 \times 10^8 / 0.61977 \times 10^{-10}$$

$$E = 3.1047 \times 10^{-15} \text{ joules}$$

$$E = 3.1047 \times 10^{-15} / 1.602 \times 10^{-19}$$

$$E = 19,947 \text{ eV}$$

Voltage = 19.947 KV

B. What is the frequency of Mo $K\beta$ radiation?

$$\nu = c/\lambda$$

$$\nu = 3 \times 10^8 / 0.63225 \times 10^{-10}$$

$$\nu = 4.745 \times 10^{18} \text{ hz}$$

C. Nb has an absorption edge corresponding to a wavelength of 0.65291 Å. Can Nb be used as a β -filter for Mo radiation? Why?

$K\beta$ (Mo) = 0.63225 Å Is this energetic enough to remove inner K-shell electrons from Nb?

Yes

$K\alpha$ (Mo) = 0.70926 Å Is this energetic enough to remove inner K-shell electrons from Nb?

No

Then the $K\beta$ of Mo will be absorbed strongly but the $K\alpha$ will not. So it can be used as a β -filter.

Planck's Constant = 6.6×10^{-34} joule-sec

1 eV = 1.6016×10^{-19} joule

c = 3.0×10^8 m/sec

GEOL3010

X-Rays

Problem Set 7

2. Barite (BaSO_4) has orthorhombic cell edges $a = 7.157 \text{ \AA}$, $b = 8.884 \text{ \AA}$, and $c = 5.457 \text{ \AA}$. Calculate 2θ for $\text{CuK}\alpha$ radiation $\lambda = 1.5405 \text{ \AA}$ for the following X-ray diffractions:

a. (002)

$$d = 1/[h^2/a^2 + k^2/b^2 + l^2/c^2]^{1/2}$$

$$d = c/2$$

$$d = 2.728 \text{ \AA}$$

$$\lambda = 2d \cdot \sin \theta$$

$$2\theta = 2 \cdot \sin^{-1}(\lambda/2d)$$

$$2\theta = 2 \cdot \sin^{-1}(1.5405/5.457)$$

$$2\theta = 32.79^\circ$$

b. (110)

$$d = 1/[h^2/a^2 + k^2/b^2 + l^2/c^2]^{1/2}$$

$$d = 1/[(1/7.157)^2 + (1/8.884)^2]^{1/2}$$

$$d = 5.574 \text{ \AA}$$

$$\lambda = 2d \cdot \sin \theta$$

$$2\theta = 2 \cdot \sin^{-1}(\lambda/2d)$$

$$2\theta = 2 \cdot \sin^{-1}(1.5405/2 \cdot 5.574)$$

$$2\theta = 15.88^\circ$$

c. (021)

$$d = 1/[h^2/a^2 + k^2/b^2 + l^2/c^2]^{1/2}$$

$$d = 1/[(2/8.884)^2 + (1/5.457)^2]^{1/2}$$

$$d = 3.445 \text{ \AA}$$

$$\lambda = 2d \cdot \sin \theta$$

$$2\theta = 2 \cdot \sin^{-1}(\lambda/2d)$$

$$2\theta = 2 \cdot \sin^{-1}(1.5405/2 \cdot 3.445)$$

$$2\theta = 25.84^\circ$$

d. (111)

$$d = 1/[h^2/a^2 + k^2/b^2 + l^2/c^2]^{1/2}$$

$$d = 1/[(1/7.157)^2 + (1/8.884)^2 + (1/5.457)^2]^{1/2}$$

$$d = 3.899 \text{ \AA}$$

$$\lambda = 2d \cdot \sin \theta$$

$$2\theta = 2 \cdot \sin^{-1}(\lambda/2d)$$

$$2\theta = 2 \cdot \sin^{-1}(1.5405/2 \cdot 3.899)$$

$$2\theta = 22.79^\circ$$

e. (301)

$$d = 1/[h^2/a^2 + k^2/b^2 + l^2/c^2]^{1/2}$$

$$d = 1/[(3/7.157)^2 + (1/5.457)^2]^{1/2}$$

$$d = 2.262 \text{ \AA}$$

$$\lambda = 2d \cdot \sin \theta$$

$$2\theta = 2 \cdot \sin^{-1}(\lambda/2d)$$

$$2\theta = 2 \cdot \sin^{-1}(1.5405/2 \cdot 2.262)$$

$$2\theta = 39.82^\circ$$