

Space Groups

Internal Symmetry of Crystals
Dyar, Gunter, Tasa chapt 12

Space Groups

- If translation operations are included with rotation and inversion,
- We have 230 three-dimensional space groups.
- Translation operations
 - Unit cell translations
 - Centering operations (Lattices) (A, B, C, I, F, R)
 - Glide planes (reflection + translation) (a, b, c, n, d)
 - Screw axes (rotation + translation) (2₁, 3₁, 3₂)

Space Groups: Learning Goals

- Our knowledge of crystals is very well established and defined.
- Crystal structures are determined by X-ray diffraction methods (+neutron & electron)
- Recognize a space group symbol.
 - Pbnm, C2/c, Ia-3d, P-1, Imma, etc.
 - Every mineral structure has one.

Space Groups

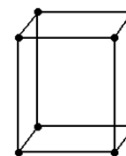
- 230 three-dimensional space groups
- Hermann-Mauguin symbols. (4 positions)
- First position is Lattice type (P, A, B, C, I, F or R)
- Second, third and fourth positions as with point groups

Space Groups: Learning Goals

- Mineral structure is defined by:
 - Space Group (class, system)
 - Unit cell parameters (a, b, c, α , β , γ).
 - Coordinates (fractional) of unique atoms

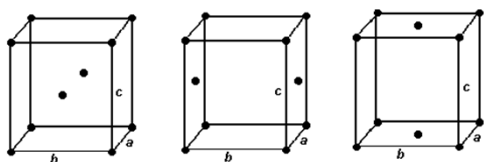
Lattices (centering operations)

- P is for primitive

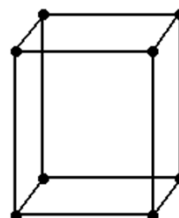


Lattices

- *A, B, and C are end-centered*



Lattices

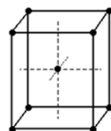


- A. *P*
- B. *B*
- C. *I*
- D. *R*
- E. *F*

Lattices

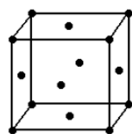
- *I is body-centered*

- *point @ (0.5, 0.5, 0.5)*
- *multiplicity = 2*

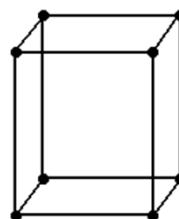


- *F is face-centered*

- *0.5, 0.5, 0*
- *0.5, 0, 0.5*
- *0, 0.5, 0.5*
- *multiplicity = 4*



Lattices

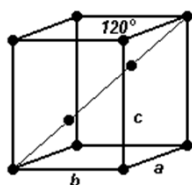


- A. *P*
- B. *B*
- C. *I*
- D. *R*
- E. *F*

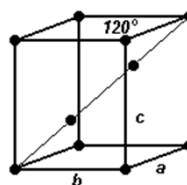
Lattices

- *R is rhombohedral*

- *(2/3, 1/3, 1/3)*
- *(1/3, 2/3, 2/3)*
- *multiplicity = 3*
- *Trigonal system*

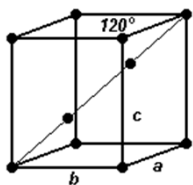


Lattices



- A. *P*
- B. *B*
- C. *I*
- D. *R*
- E. *F*

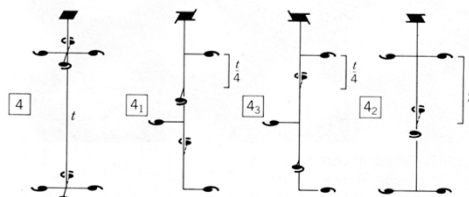
Lattices



- A. *P*
- B. *B*
- C. *I*
- D. *R*
- E. *F*

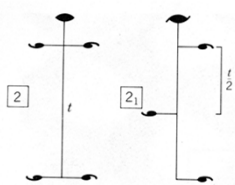
Screw Tetrads

- 4_1 is a 90° rotation plus a $1/4$ cell translation (right-handed).
- 4_2 is a 90° rotation plus a $1/2$ cell translation (no handedness).
- 4_3 is a 90° rotation plus a $3/4$ cell translation (left-handed).



Rotation + Translation = Screw Axes Screw Diads

- 2 is a 180° rotation
- 2_1 is a 180° rotation plus $1/2$ cell translation.

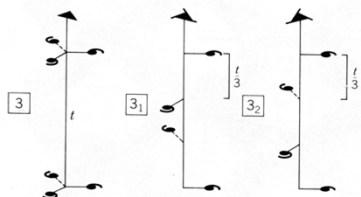


Screw Hexads

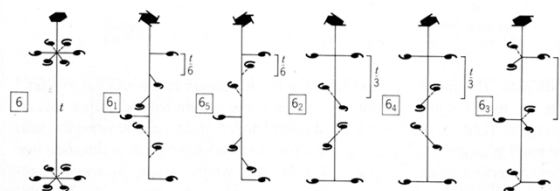
- 6_1 is a 60° rotation plus a $1/6$ cell translation (right-handed).
- 6_2 is a 120° rotation plus a $1/3$ cell translation (right-handed).
- 6_3 is a 180° rotation plus a $1/2$ cell translation (no handedness).
- 6_4 is a 240° rotation plus a $2/3$ cell translation (left-handed).
- 6_5 is a 300° rotation plus a $5/6$ cell translation (left-handed).

Screw Triads

- 3_1 is a 120° rotation plus a $1/3$ cell translation.
- 3_2 is a 120° rotation plus a $2/3$ cell translation



Screw Hexads



Glide Planes = reflection + translation

$\begin{array}{cccccccc} \square & & \square & & \square & & \square & & \square \\ \hline & \square & & \square & & \square & & \square & \end{array}$

Space Group Symmetry Diagrams

	Axis	Plane
- a	2 ₁	b
- b	2	c
- c	2 ₁	n

- $P2_1|b2/c2_1|n = Pbcn$

Glide Planes

- Reflection plus 1/2 cell translation
 - a-glide: a/2 translation
 - b-glide: b/2 translation
 - c-glide: c/2 translation
 - n-glide(normal to a): b/2+c/2 translation
 - n-glide(normal to b): a/2+c/2 translation
 - n-glide(normal to c): a/2+b/2 translation
 - d-glide (tetragonal + cubic systems)

Space Group Symmetry Diagrams

	Axis	Plane
- a	2	b
- b	2	a
- c	2	n

- $P2|b2/a2|n = Pban$

Space Group Symmetry Diagrams

- a-vertical
- b-horizontal
- c-normal to page

Perpendicular to a:

- A. n-glide
- B. b-glide
- C. c-glide
- D. Mirror
- E. nothing

Perpendicular to a:

• D. Mirror

Perpendicular to c:

- A. n-glide
- B. b-glide
- C. c-glide
- D. Mirror
- E. nothing

Perpendicular to b:

- A. n-glide
- B. b-glide
- C. c-glide
- D. Mirror
- E. nothing

Perpendicular to c:

- Mirror

Perpendicular to b:

- c-glide

Axis Parallel to a:

- A. 2
- B. 2₁
- C. 4
- D. nothing

Axis Parallel to a:

▸ A. 2
 ▸ B. 2_1
 ▸ C. 4
 ▸ D. nothing

Primitive space group

▸ $P 2/m 2/c 2_1/m$
 ▸ = $Pmcm$

Axis Parallel to c:

▸ A. 2
 ▸ B. 2_1
 ▸ C. 4
 ▸ D. nothing

Pbcn

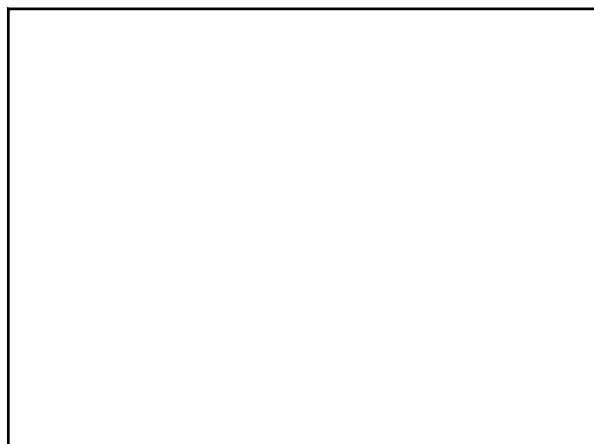
- (x, y, z)
- We place an atom at fractional coordinates x, y, z in the unit cell.
- The symmetry operators repeat that atom 7 times to give 8 identical atoms in the cell.

Axis Parallel to c:

▸ A. 2
 ▸ B. 2_1
 ▸ C. 4
 ▸ D. nothing

Pbcn

- (x, y, z)
- $(-x, -y, -z)$
- $(x, -y, 1/2+z)$
- $(-x, y, 1/2-z)$
- $(1/2 - x, 1/2 + y, z)$
- $(1/2 + x, 1/2 - y, -z)$
- $(1/2 - x, 1/2 - y, 1/2 + z)$
- $(1/2 + x, 1/2 + y, 1/2 - z)$

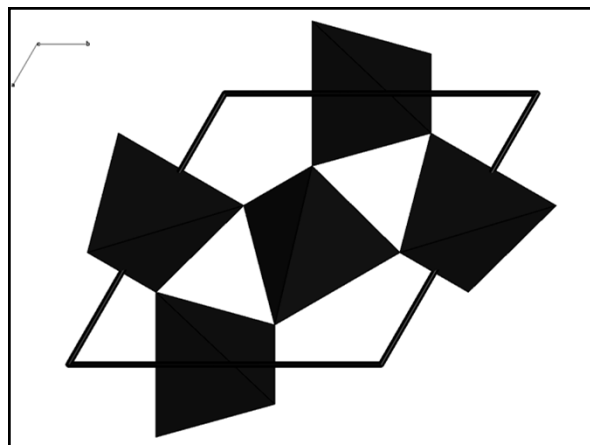


Quartz $P3_121$

- Cell $a=4.9137$; $b=4.9137$ $c= 5.4047\text{\AA}$
- Space Group: $P3_121$
- Si $(0.4697, 0.0, 0.0)$
- O $(0.4133, 0.2672, 0.1188)$

Enstatite $Pbcn$

- Cell $a=9.306$; $b=8.892$ $c= 5.349\text{\AA}$
 - What's in the cell??
 - Mg1 special 4
 - Mg2 special 4
 - Si general 8
 - O1 general 8
 - O2 general 8
 - O3 general 8
- _____ 8 Mg, 8 Si, 24 O



Enstatite $Pbcn$

- Cell $a=9.306$; $b=8.892$ $c= 5.349\text{\AA}$
 - What's in the cell??
- _____ 8 Mg, 8 Si, 24 O
- 8 x (MgSiO_3)
 - $Z = 8$ (formula units per cell)

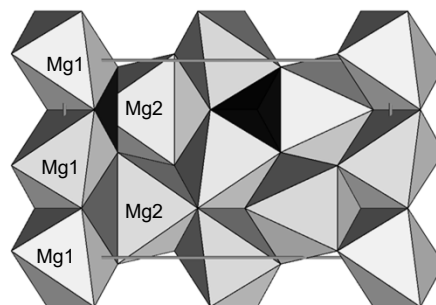
Quartz $P3_121$

- Cell $a=4.9137$; $b=4.9137$ $c= 5.4047\text{\AA}$
- Space Group: $P3_121$ or $P3_221$
- Si (3)
- O (6)
- Formula 3 x (SiO_2)

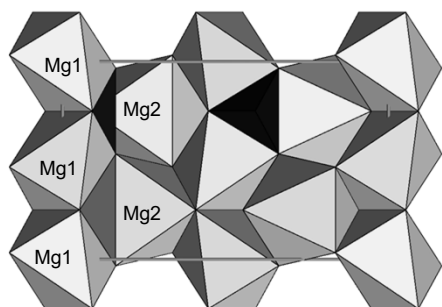
Olivine $Pbnm$ Mg_2SiO_4

- Cell $a=4.756$; $b=10.207$ $c= 5.980 \text{ \AA}$
- Mg1 4 (0.0, 0.0, 0.0)
- Mg2 4 (0.991, 0.277, 0.25)
- Si 4 (0.426, 0.094, 0.25)
- O1 4 (0.766, 0.091, 0.25)
- O2 4 (0.222, 0.447, 0.25)
- O3 8 (0.278, 0.163, 0.033)

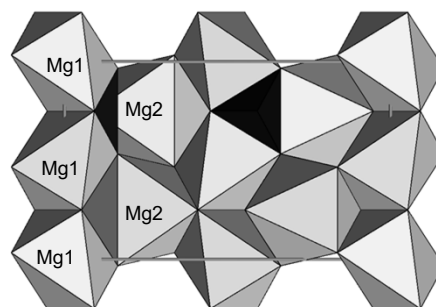
Cation Substitution Olivine



Olivine $Pbnm$ Mg_2SiO_4



Fe^{2+} substitutes for Mg^{2+} in octahedral sites



Olivine $Pbnm$ Mg_2SiO_4

- Cell $a=4.756$; $b=10.207$ $c= 5.980 \text{ \AA}$
- Mg1 4, Mg2 4, Si 4
- O1 4, O2 4, O3 8
- = $4 \times (Mg_2SiO_4)$
- This is why minerals have formulas with integer atom ratios.

Cation Substitution

- Mg_2SiO_4 - Fe_2SiO_4 Olivine
- Mg_2SiO_4 Forsterite
- Fe_2SiO_4 Fayalite
-
- $(Mg,Fe)_2SiO_4$ Any intermediate composition
- 2 divalent, 1 tetravalent, 4 oxygens

Cation Substitution

- Mg_2SiO_4 - Fe_2SiO_4 Olivine
- Mg_2SiO_4 Forsterite Fo
- Fe_2SiO_4 Fayalite Fa
-
- $Fo_{90}Fa_{10}$

Lab Next Week: XTALDRAW

- *Bring: Your mineral*
 - *Unit Cell information*
 - *a, b, c, α , β , γ*
 - *Formula*
 - *Atom positions*